

# Comparative Study of Bismuth Tellurites by Infrared Absorption Spectroscopy

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**Infrared spectra of  $\alpha$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub>,  $\beta$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub>, Bi<sub>2</sub>Te<sub>2</sub>O<sub>7</sub>, and Bi<sub>2</sub>TeO<sub>5</sub>, have been compared and analyzed. The bands in the region of 620–400 cm<sup>-1</sup> correspond to the stretching modes of the bismuth–oxygen framework, while those in the region of 800–620 cm<sup>-1</sup> belong to the stretching vibrations of the Te–O bonds. TeO<sub>3</sub> polyhedra of trigonal–pyramidal configuration with equal bond lengths (within 0.02 nm) have been found in bismuth tellurites. The largest deformation of TeO<sub>3</sub> has been observed in Bi<sub>2</sub>TeO<sub>5</sub> as compared to the other bismuth tellurites investigated in this study. Connected TeO<sub>3</sub> polyhedra form Te<sub>2</sub>O<sub>5</sub> groups in the structure of  $\alpha$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub>. The relation between the frequency of the asymmetric stretching mode and the length of Bi–O bond is linear for Bi<sub>2</sub>TeO<sub>5</sub> and nonlinear for  $\alpha$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub> and  $\alpha$ -Bi<sub>2</sub>O<sub>3</sub>. © 2000 Academic Press**

**Key Words:** bismuth tellurium oxide; IR-spectra.

## INTRODUCTION

Bismuth-oxide-based compounds have a number of important applications. The  $\delta$ -Bi<sub>2</sub>O<sub>3</sub>-type compounds have anion-deficient fluorite structures and are applied as oxygen ion conductors (1, 2). The bismuth tellurium oxides such as  $\beta$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub>, Bi<sub>2</sub>Te<sub>2</sub>O<sub>7</sub>, and Bi<sub>2</sub>TeO<sub>5</sub> structurally belong to the  $\delta$ -Bi<sub>2</sub>O<sub>3</sub>-type compounds (3, 4). Among them the most important is Bi<sub>2</sub>TeO<sub>5</sub>, which shows peculiar photorefractive properties (5). Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub> has two modifications:  $\beta$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub> belongs to the  $\delta$ -Bi<sub>2</sub>O<sub>3</sub> type compounds while  $\alpha$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub> is similar to  $\alpha$ -Bi<sub>2</sub>O<sub>3</sub> (3). The monoclinic  $\alpha$ -Bi<sub>2</sub>O<sub>3</sub> is stable up to 729°C and above this temperature it transforms to  $\delta$ -Bi<sub>2</sub>O<sub>3</sub>. On cooling  $\delta$ -Bi<sub>2</sub>O<sub>3</sub>, two metastable modifications may occur, i.e., the tetragonal  $\beta$ -Bi<sub>2</sub>O<sub>3</sub> and the bcc  $\gamma$ -Bi<sub>2</sub>O<sub>3</sub> phase. The  $\gamma$ -Bi<sub>2</sub>O<sub>3</sub>-type compounds are called sillenites and can be written by the formula of Bi<sub>12</sub>MO<sub>20</sub> (*M* = Ge, Si, Ti). Due to the piezoelectric, elec-

trooptical, and elasto-optical properties they have many technological applications (6).

The aim of this work is to characterize bismuth tellurite compounds by infrared spectroscopy and compare the results with other structural investigations. As a starting point the IR-spectra of  $\alpha$ -Bi<sub>2</sub>O<sub>3</sub> and  $\alpha$ -TeO<sub>2</sub> are revisited.

The monoclinic alpha Bi<sub>2</sub>O<sub>3</sub> contains two sorts of bismuth atoms. One of them has five oxygen neighbors at the corners of a distorted octahedron with the inert pair of electrons occupying the sixth corner (the lengths of Bi–O bonds are placed between 0.21 and 0.26 nm). Two more oxygen atoms are weakly bonded (0.32 nm, 0.34 nm). The other sort of bismuth has six oxygen neighbors at the corners of a distorted octahedron (lengths of Bi–O bonds are placed between 0.21 and 0.28 nm) (7, 8). An intense broad band consisting of six components was reported between 595 and 380 cm<sup>-1</sup> in the IR spectrum of  $\alpha$ -Bi<sub>2</sub>O<sub>3</sub> (9). Because of compounds containing Bi<sub>2</sub>O<sub>3</sub> have similar IR spectra in this region, it was concluded that these bands are predominantly connected to the stretching modes of bismuth–oxygen framework (9, 10). In sillenites for instance the frequencies of asymmetric stretching modes of Bi–O bonds were observed in the range of 620–450 cm<sup>-1</sup>. The relation between the frequencies of the four modes and the Bi–O bond lengths was found to be approximately linear for sillenites and nonlinear for  $\alpha$ -Bi<sub>2</sub>O<sub>3</sub>.

The  $\alpha$ -TeO<sub>2</sub> crystallizes in distorted rutile structure. Tellurium has distorted trigonal bipyramidal fourfold coordination by oxygen with two equatorial (0.1878 nm) and two axial bonds (0.2122 nm) (11, 12). In the IR spectrum of paratellurite two intense bands were observed at 675 and 780 cm<sup>-1</sup>, having two shoulders (635, 714 cm<sup>-1</sup>) at the first band (13). The four bands correspond to the stretching modes of the equatorial and axial bonds:  $\nu_{\text{eq}}^{\text{s}} > \nu_{\text{eq}}^{\text{as}} > \nu_{\text{ax}}^{\text{as}} > \nu_{\text{ax}}^{\text{s}}$ . The  $\alpha$ -TeO<sub>2</sub> is built up by TeO<sub>4</sub> groups. However,

**TABLE 1**  
**Lattice Types and Space Groups**

$\alpha$ -Bi <sub>2</sub> O <sub>3</sub>	$\alpha$ -Bi <sub>2</sub> Te <sub>4</sub> O <sub>11</sub>	$\delta$ -Bi <sub>2</sub> O <sub>3</sub>	$\beta$ -Bi <sub>2</sub> Te <sub>4</sub> O <sub>11</sub>	Bi <sub>2</sub> Te <sub>2</sub> O <sub>7</sub>	Bi <sub>2</sub> TeO <sub>5</sub>	$\alpha$ -TeO <sub>2</sub>
(7, 8, 16)	(3, 17, 18)	(7, 16)	(3, 18)	(19)	(4, 20)	(11, 12)
Monoclinic	Monoclinic	Cubic	Cubic	Orthorhombic	Orthorhombic	Tetragonal
<i>P2<sub>1</sub>/c</i>	<i>P2<sub>1</sub>/n</i>	<i>Fm3m</i>	<i>Fm3m</i>	<i>Pcnb</i>	<i>Abm2</i>	<i>P4<sub>1</sub>2<sub>1</sub>2</i>

some tellurites contain TeO<sub>3</sub> polyhedra in trigonal-pyramidal configuration with equal lengths of Te-O bonds. Two bands of symmetric and degenerate TeO<sub>3</sub> modes ( $\nu_{\text{TeO}_3}^{\text{s}}$  and  $\nu_{\text{TeO}_3}^{\text{d}}$ ) were observed in the IR spectra of these tellurites. In certain tellurites the bands split as a consequence of symmetry loss in the TeO<sub>3</sub> group (13).

IR spectra of Bi<sub>2</sub>TeO<sub>5</sub> have been published by Földvári *et al.* (14). They investigated single crystal samples with thicknesses of 0.095 and 16 mm in the ranges of 1600–850 and 2000–1600 cm<sup>-1</sup>, respectively, at RT and 10 K. Kucha *et al.* (15) also published absorption spectra of Bi<sub>2</sub>TeO<sub>5</sub> on single crystal samples of 0.43 and 1.3 mm thickness in the ranges of 2200–400 and 4000–2000 cm<sup>-1</sup> and found at about 40 absorption bands. The bands at 425, 465, 500, 540, 570, and 610 cm<sup>-1</sup> were assigned to the vibrations of Bi–O bonds. The bands at 670 and 705 cm<sup>-1</sup> were assumed to belong to the Te–O bond. The bands at higher frequencies were assigned to the second harmonics and combinations of the low frequency bands.

In this paper we present IR absorption spectra of powdered  $\alpha$ -Bi<sub>2</sub>O<sub>3</sub>,  $\alpha$ -TeO<sub>2</sub>,  $\alpha$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub>,  $\beta$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub>, Bi<sub>2</sub>Te<sub>2</sub>O<sub>7</sub>, and Bi<sub>2</sub>TeO<sub>5</sub>. For convenience the available literature data on lattice types, space groups, and interatomic distances are given in Tables 1–3.

## EXPERIMENTAL

Bismuth oxide (Bi<sub>2</sub>O<sub>3</sub>) from Johnson Matthey (Grade 1) and tellurium oxide (TeO<sub>2</sub>) prepared from tellur (99.999%) by oxidation with double distilled HNO<sub>3</sub> (Carlo Erba for

analysis) were used as starting materials. These materials were heated up to 450°C in air and after cooling they were ground and sieved. Powders with less than 63  $\mu\text{m}$  particle size were used for the synthesis. Bi<sub>2</sub>TeO<sub>5</sub> (a), Bi<sub>2</sub>Te<sub>2</sub>O<sub>7</sub> (b),  $\beta$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub> (c), and  $\alpha$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub> (d) were prepared by the following procedures:

(a) Bi<sub>2</sub>O<sub>3</sub> and TeO<sub>2</sub> powders were mixed in the ratio of 1:1 and fired at 850°C for 120 min under argon atmosphere in a platinum crucible.

(b) Bi<sub>2</sub>O<sub>3</sub> and TeO<sub>2</sub> powders in the ratio of 1:2 were treated exactly in the same way, but the firing temperature was 700°C.

(c) Bi<sub>2</sub>O<sub>3</sub> and TeO<sub>2</sub> powders in the ratio of 1:4 were treated as previous samples, except the firing was employed at 650°C for 30 min.

(d)  $\beta$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub> was annealed at 520°C under argon atmosphere for 10 min.

The composition of the samples was checked by X-ray diffraction using a Philips PW 1730 X-ray generator with CuK $\alpha$  source. For IR measurements pellets were prepared by mixing the investigated compounds with KBr. The IR spectra were taken by a JASCO FT/IR-300E spectrometer at room temperature. The absorption bands were studied in the range of 1000–400 cm<sup>-1</sup>.

## RESULTS AND DISCUSSION

Figure 1 shows the IR spectra of  $\beta$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub>, Bi<sub>2</sub>Te<sub>2</sub>O<sub>7</sub>,  $\alpha$ -Bi<sub>2</sub>Te<sub>4</sub>O<sub>11</sub>, Bi<sub>2</sub>TeO<sub>5</sub>,  $\alpha$ -Bi<sub>2</sub>O<sub>3</sub>, and  $\alpha$ -TeO<sub>2</sub>. Vibrational frequencies of the stretching modes of Te–O and Bi–O

**TABLE 2**  
**Interatomic Te–O Distances (nm) in Tellurites**

$\alpha$ -Bi <sub>2</sub> Te <sub>4</sub> O <sub>11</sub> (17)				Bi <sub>2</sub> TeO <sub>5</sub> (20) Te–O
Te1–O	Te2–O	Te3–O	Te4–O	
0.187	0.187	0.187	0.187	0.176
0.188	0.190	0.190	0.189	0.186
0.193	0.197	0.207	0.196	0.195
0.271	0.25	0.228	0.241	0.307
0.296	0.274	0.277	0.295	0.324
0.298	0.298	0.328	0.299	
0.343	0.301	0.336	0.322	
	0.343			

**TABLE 3**  
**Interatomic Bi–O Distances (nm) in Tellurites**

$\alpha$ -Bi <sub>2</sub> Te <sub>4</sub> O <sub>11</sub> (17)		Bi <sub>2</sub> TeO <sub>5</sub> (20)		
Bi1–O	Bi2–O	Bi1–O	Bi2–O	Bi3–O
0.220	0.222	0.210	0.219	0.196
0.224	0.233	0.231	0.227	0.205
0.231	0.235	0.231	0.227	0.205
0.239	0.240	0.244	0.242	0.286
0.250	0.247	0.261	0.242	0.286
0.278	0.267	0.268	0.302	0.305
0.279	0.284	0.304	0.302	0.305
	0.293	0.305		

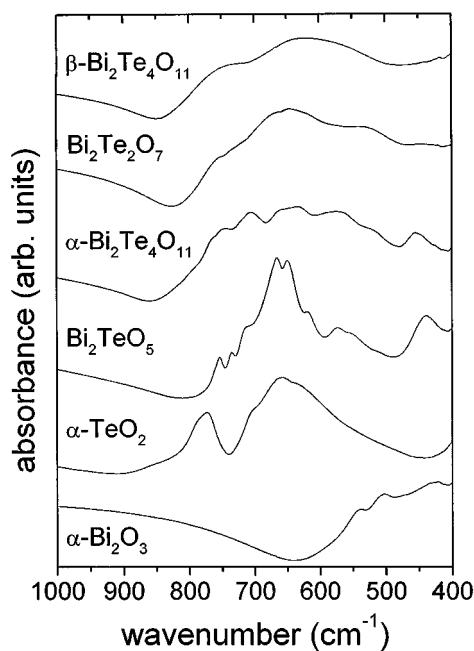


FIG. 1. Absorption spectra of bismuth tellurites,  $\text{TeO}_2$ , and  $\alpha\text{-Bi}_2\text{O}_3$ .

bonds in various bismuth tellurites are summarized in Table 4. The band positions were determined by the help of the second derivatives of the spectra. Four bands can be seen in the IR spectrum of our  $\alpha\text{-TeO}_2$  sample in Fig. 1 at 770, 708, 666, and 632  $\text{cm}^{-1}$ . These bands, however, are shifted by 5–10  $\text{cm}^{-1}$  toward lower frequencies as compared to Ref. (13). Spectra of bismuth tellurium oxides are similar to each other, but differ from the spectrum of  $\text{TeO}_2$  because their structures consist of  $\text{TeO}_3$  groups. IR spectra of bismuth tellurites generated by  $\text{TeO}_3$  groups can be characterized by two vibrations of symmetric and degenerate  $\text{TeO}_3$  modes of the  $\text{Te-O}$  bonds. The two ranges are 765–730 and 670–630  $\text{cm}^{-1}$ , respectively. Splitting of these bands is observed in the case of  $\text{Bi}_2\text{Te}_2\text{O}_7$ ,  $\alpha\text{-Bi}_2\text{Te}_4\text{O}_{11}$ , and  $\text{Bi}_2\text{TeO}_5$  due to decrease in the symmetry of  $\text{TeO}_3$  group. In Fig. 1 one can observe the similarity between the spectra of  $\beta\text{-Bi}_2\text{Te}_4\text{O}_{11}$

and  $\text{Bi}_2\text{Te}_2\text{O}_7$  in peak positions and shape. They have strong overlapping peaks.

#### $\beta\text{-Bi}_2\text{Te}_4\text{O}_{11}$

In the spectrum of  $\beta\text{-Bi}_2\text{Te}_4\text{O}_{11}$  symmetric and degenerate  $\text{TeO}_3$  modes of the  $\text{Te-O}$  bonds can be found at  $\nu_{\text{TeO}_3}^s = 765 \text{ cm}^{-1}$  and  $\nu_{\text{TeO}_3}^d = 627 \text{ cm}^{-1}$ . Overlapping bands in the 600 to 400  $\text{cm}^{-1}$  wavenumber range correspond to the asymmetric stretching mode of the  $\text{Bi-O}$  bond. Since no  $\text{Bi-O}$  bond lengths are available, no relation between the bond lengths and vibrational frequency can be established.

#### $\text{Bi}_2\text{Te}_2\text{O}_7$

In the spectrum of  $\text{Bi}_2\text{Te}_2\text{O}_7$  a splitting of the vibrations of degenerate  $\text{TeO}_3$  modes was observed ( $\nu_{\text{TeO}_1} = 757 \text{ cm}^{-1}$ ,  $\nu_{\text{TeO}_2} = 670 \text{ cm}^{-1}$  and  $\nu_{\text{TeO}_3} = 647 \text{ cm}^{-1}$ ) suggesting that the  $\text{TeO}_3$  group has lower symmetry in  $\text{Bi}_2\text{Te}_2\text{O}_7$  than in  $\beta\text{-Bi}_2\text{Te}_4\text{O}_{11}$ . Several overlapping bands can be detected in the range of low frequencies. Similarly to  $\beta\text{-Bi}_2\text{Te}_4\text{O}_{11}$ , no assignable interatomic  $\text{Bi-O}$  distances are available.

#### $\alpha\text{-Bi}_2\text{Te}_4\text{O}_{11}$

In the region of 800–620  $\text{cm}^{-1}$  there is a new band in the spectrum of  $\alpha\text{-Bi}_2\text{Te}_4\text{O}_{11}$  compared to that of  $\text{Bi}_2\text{Te}_2\text{O}_7$ . It is assumed that a  $\text{Te}_2\text{O}_5$  group is formed from two  $\text{TeO}_3$  polyhedra sharing an oxygen. Four stretching modes of the  $\text{Te-O}$  bonds appear. We assign two of them to  $\text{TeO}_2$  groups and the two others to the  $\text{Te-O-Te}$  bridge, similarly to the compounds  $\text{MgTe}_2\text{O}_5$  (13, 21) and  $\text{Te}_2\text{V}_2\text{O}_9$  (13, 22). The vibrational frequencies are  $\nu_{\text{TeO}_2}^s = 749 \text{ cm}^{-1}$ ,  $\nu_{\text{TeO}_2}^{\text{as}} = 706 \text{ cm}^{-1}$ ,  $\nu_{\text{Te-O-Te}}^{\text{as}} = 667 \text{ cm}^{-1}$ , and  $\nu_{\text{Te-O-Te}}^s = 631 \text{ cm}^{-1}$ . This result is consistent with Refs. (17, 18), where it was established that the structure of  $\alpha\text{-Bi}_2\text{Te}_4\text{O}_{11}$  consisted of an ordered intergrowth of fluorite  $\text{Bi}_2\text{Te}_2\text{O}_7$  and rutile or

TABLE 4  
Vibrational Frequencies of the Stretching Modes of  $\text{Te-O}$  and  $\text{Bi-O}$  Bonds in Bismuth Tellurites

Compounds	Vibrations of $\text{Te-O}$ bonds ( $\text{cm}^{-1}$ )				Vibrations of $\text{Bi-O}$ bonds ( $\text{cm}^{-1}$ )						
$\beta\text{-Bi}_2\text{Te}_4\text{O}_{11}$	765			627							419
$\text{Bi}_2\text{Te}_2\text{O}_7$	757		670	647				528	450		427
$\alpha\text{-Bi}_2\text{Te}_4\text{O}_{11}$	749		706	667	631	589	562	510	495	457	425
$\text{Bi}_2\text{TeO}_5$	755	737	716	667	647	616	575	549	509	441	
$\alpha\text{-Bi}_2\text{O}_3$						597	543	504	476	436	421
$\text{TeO}_2$	770	708	666	632							

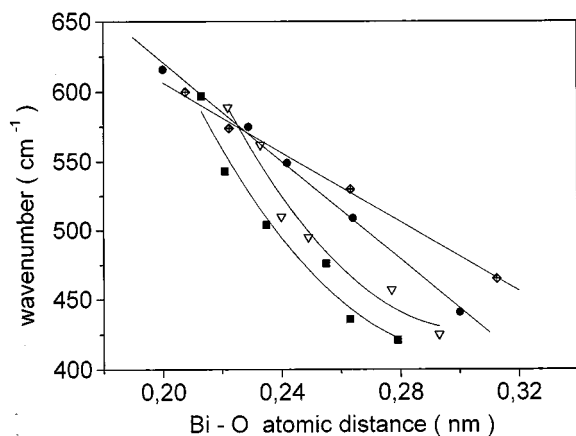
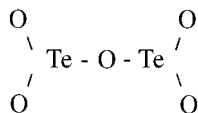


FIG. 2. Relation between vibrational frequencies and groups of Bi-O bond lengths in bismuth tellurites.  $\alpha$ - $\text{Bi}_2\text{O}_3$  (squares),  $\alpha$ - $\text{Bi}_2\text{Te}_4\text{O}_{11}$  (triangles),  $\text{Bi}_2\text{TeO}_5$  (circles), and  $\text{Bi}_{12}\text{GeO}_{20}$  (diamonds).  $\text{Bi}_{12}\text{GeO}_{20}$  data are reproduced from R. J. Betsch and W. B. White, *Spectrochim. Acta A* **34**, 505 copyright 1977, with permission from Elsevier Science.

fluorite  $\text{TeO}_2$  slabs containing



subunits. Six asymmetric stretching modes of Bi-O bonds are at 589, 562, 510, 495, 457, and 425  $\text{cm}^{-1}$ . These six vibrational frequencies correspond to six groups of bond lengths. The group can be defined as an average value of bond lengths differing by less than 0.013 nm. The relation between the vibrational frequencies and groups of Bi-O bond lengths is shown in Fig. 2. This relation is found, similarly to the case of  $\alpha$ - $\text{Bi}_2\text{O}_3$ , to be nonlinear.

### $\text{Bi}_2\text{TeO}_5$

The largest number of bands were observed in the spectrum of  $\text{Bi}_2\text{TeO}_5$ . The five highest energy bands belong to Te-O bonds. From the biggest splitting of the degenerate  $\text{TeO}_3$  mode the largest deformation of the  $\text{TeO}_3$  polyhedra can be concluded. The difference (0.02 nm) between the three shortest Te-O bonds is the biggest in our investigated bismuth tellurites. This value is rather high even in comparison with other tellurites containing tellurium atoms with threefold coordination. In the range of low frequencies there are five asymmetric stretching modes of the Bi-O framework at 616, 575, 549, 509, and 441  $\text{cm}^{-1}$ . Comparing our spectrum with that of Kucha *et al.* (15) there are differences in both the region of 1000–620  $\text{cm}^{-1}$  and the region of 620–400  $\text{cm}^{-1}$ . In the present work we used sintered powders of stoichiometric  $\text{Bi}_2\text{TeO}_5$ , while Kucha *et al.* used

single crystal samples. Földvári *et al.* (14) have reported on the strong nonstoichiometric character of the  $\text{Bi}_2\text{TeO}_5$  crystal. In Fig. 2 the relation between the frequencies of asymmetric stretching modes and groups of Bi-O bond lengths in  $\text{Bi}_2\text{TeO}_5$  is shown. The dependence for  $\text{Bi}_2\text{TeO}_5$  is linear, as seen in sillenites (9) and in contrast to that seen in  $\alpha$ - $\text{Bi}_2\text{Te}_4\text{O}_{11}$  and  $\alpha$ - $\text{Bi}_2\text{O}_3$ . This fact can be interpreted on a structural basis. In  $\text{Bi}_2\text{TeO}_5$  the Bi2-O and Bi3-O polyhedra look very similar to the Bi-O polyhedron sillenites having Bi-O of bond lengths in three equal pairs and one short bond. Among the three pairs the two shortest are in the same plane in both compounds. The short bond and the long pair are located on the opposite side of this plane. In  $\alpha$ - $\text{Bi}_2\text{Te}_4\text{O}_{11}$ , however, the Bi-O polyhedra are more complex, as seen in the highly distorted octahedra of  $\alpha$ - $\text{Bi}_2\text{O}_3$ .

### CONCLUSION

In this paper infrared spectra of  $\alpha$ - $\text{Bi}_2\text{Te}_4\text{O}_{11}$ ,  $\beta$ - $\text{Bi}_2\text{Te}_4\text{O}_{11}$ ,  $\text{Bi}_2\text{Te}_2\text{O}_7$ , and  $\text{Bi}_2\text{TeO}_5$  have been compared and analyzed. Absorption bands were studied in the range of 1000–400  $\text{cm}^{-1}$ . Bands in the region of 620–400  $\text{cm}^{-1}$  correspond to the stretching modes of the bismuth-oxygen framework, while those in the region of 800–620  $\text{cm}^{-1}$  belong to stretching vibrations of the Te-O bonds. Tellurites contain trigonal-pyramidal configuration of  $\text{TeO}_3$  polyhedra with equal bond lengths (within 0.02 nm). Similarity can be observed between the IR spectra of  $\beta$ - $\text{Bi}_2\text{Te}_4\text{O}_{11}$  and  $\text{Bi}_2\text{Te}_2\text{O}_7$ .  $\text{Te}_2\text{O}_5$  groups are formed by connected  $\text{TeO}_3$  polyhedra in  $\alpha$ - $\text{Bi}_2\text{Te}_4\text{O}_{11}$ . Largest deformation of the  $\text{TeO}_3$  polyhedra can be concluded in  $\text{Bi}_2\text{TeO}_5$ . The difference between its three shortest Te-O bond lengths is the biggest among the investigated bismuth tellurites and moreover this value is rather high even in comparison with other tellurites containing tellurium atoms with threefold coordination. The relation between the observed IR bands and the Bi-O bond lengths has been found to be nonlinear for  $\alpha$ - $\text{Bi}_2\text{O}_3$  and  $\alpha$ - $\text{Bi}_2\text{Te}_4\text{O}_{11}$  and linear for  $\text{Bi}_2\text{TeO}_5$ .

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